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**P.E.S. College of Engineering, Mandya - 571 401**  
*(An Autonomous Institution affiliated to VTU, Belgaum)*  
**Fourth Semester, B.E., - Industrial and Production Engineering**  
**Semester End Examination; June - 2016**  
**Engineering Thermodynamics**

Time: 3 hrs

Max. Marks: 100

*Note: i) Answer FIVE full questions, selecting ONE full question from each unit.  
 ii) Assume missing data suitably.*

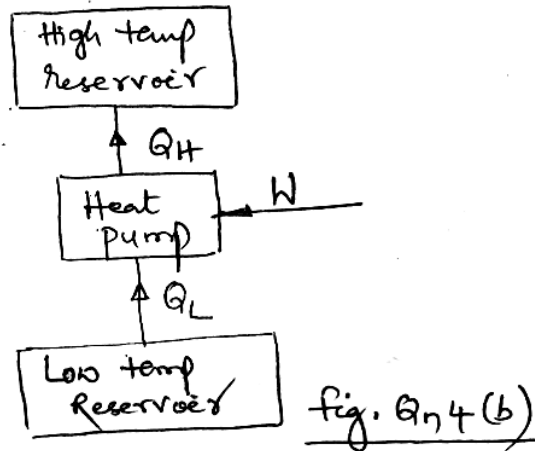
**UNIT - I**

- 1 a. Represent the following on a p-v diagram :
  - (i) Quasistatic process, indicate also the work done during this process by shading 6
  - (ii) Non-Quasistatic process 6
  - (iii) A cycle. 8
- b. Derive an expression for the work done during a Quasistatic process governed by the equations of the form  $Pv = C$  with usual rotating. 6
- c. A platinum wire is used as a resistance thermometer. The wire resistance was found be 10 ohm and 16 ohm at ice point and steam point respectively and 30 ohm at the sulphur boiling point of 444.6°C. Find the resistance of the wire at 500°C, if the resistance varies with the temperature by the relations  $R = R_0(1 + \alpha T + \beta T^2)$ . 8
- 2 a. With suitable examples, distinguish between : 8
  - (i) System and Control volume    (ii) Path functions and point functions    (iii) Process and cycle.
- b. Represent the following Quasistatic process on p-v diagram. Write expression for PdV work for these; also indicate the PdV work done during these processes by shading, 6
  - (i) Process for which  $PV = \text{constant}$     (ii) Process in which  $PV^n = \text{constant}$ .
- c. A certain fluid at 10 bar is contained in a cylinder behind a piston, the initial volume being 0.05 m<sup>3</sup>. Calculate the work done by the fluid when it expands reversibly, 6
  - (i) At constant pressure to a final volume of 0.2 m<sup>3</sup>
  - (ii) According to a law  $PV = \text{constant}$  to a final volume of 0.1 m<sup>3</sup>.

**UNIT - II**

- 3 a. Write the first law of Thermodynamics as applied to a ; 7
  - (i) Cycle    (ii) A process    (iii) A flow process and explain the terms.
- b. Represent Carnot cycle on P-V and T- S diagrams and indicate the various processes. 5
- c. A Carnot engine operates between two reservoirs at temperatures of T<sub>1</sub> and T<sub>2</sub> Kelvin. The work output of the engine is 0.6 times the heat rejected. Given that the difference is temperature between the source and the sink is 200 K, calculate; 8
  - (i) The source temperature    (ii) The sink temperature    (iii) Efficiency of the engine.

- 4 a. Derive an expression for the first law of thermodynamics as applied to a process involving a change of state and hence prove that energy is a property. 8
- b. Fig. Q<sub>n</sub>4 (b) represents a heat pump operating between two thermal reservoirs. Write an expression for the COP of the same. If this heat pump is reserved to a function as refrigeration's plant, what will be the COP term?



- c. A heat engine drives a refrigerator whose COP is 4.5. If the efficiency of the heat engine is 35% and 1260 kJ of heat is removed per hour by the refrigerator from the cold body, find the rate of heat supplied per hour to the heat engine. 8

### UNIT - III

- 5 a. Define the following : 8
- (i) Pure substance                      (ii) Saturations states  
 (iii) Critical parameters              (iv) Dryness fraction
- b. Show that for a reversible adiabatic process extended by an ideal gas pressure and volume are related by  $PV^n = \text{constant}$ . 6
- c. A vessel contains 3 kg of water at 200°C. If the volume of the vessel is 0.3 m<sup>3</sup>. Determine the, 6
- (i) Pressure    (ii) Quality    (iii) Specific enthalpy of the water vapour in the vessel.
- 6 a. On any isobar represented on a T-S diagram for water, indicate the following conditions : 8
- (i) Sub cooled liquid                      (ii) Saturated liquid                      (iii) Saturated vapour  
 (iv) Super heated vapour                      (v) Wet vapour
- b. Show that for an ideal gas  $C_p - C_v = R$ . 6
- c. Steam at 40 bar has a specific volume 0.028 m<sup>3</sup>/kg. What is the condition of steam? Find; 6
- (i) Temperature of steam    (ii) Specific internal energy    (iii) Specific enthalpy.

**UNIT - IV**

- 7 a. For a given  $T_2$ , show that the Rankine cycle efficiency depends on the mean temperature of heat additions. 6
- b. Representing the Otto cycle on a T-S diagram, derive an expression for air standard efficiency of the same with usual notations. 6
- c. An ideal diesel engine has a diameter of 15 cm and stroke of 20 cm, the clearance volume is 10% of the swept volume. Determine the compression ratio and the air standard efficiency of the engine if the cut-off takes place at 6% of the stroke. 8
- 8 a. When is reheating of steam is recommended in a steam power plant? Represent a single stage reheat Rankine cycle on a T-S diagram and write expressions for efficiency and steam rate. 8
- b. Represent Otto and diesel cycles on P-V and T- S diagrams and name the different processes. 6
- c. An engine equipped with a cylinder having a bore of 15 cm and a stroke of 45 cm operates on Otto cycle. If the clearance volume is  $2000 \text{ cm}^3$ , compute the air standard efficiency. 6

**UNIT - V**

- 9 a. Show that the efficiency of the Brayton cycle depends only on the pressure ratio; also represent the cycle on P-V and T- S diagrams. 8
- b. Draw the P-V diagram for a two stage compressor with perfect inter cooling between states. Show the saving in work done by shading; also represent the isothermal line. 6
- c. A single stage reciprocating compressor takes  $1 \text{ m}^3$  of air per minute at 1.013 bar and  $15^\circ\text{C}$  and delivers it at 7 bar. Assuming that the law of compression is  $PV^{1.35} = C$  and that clearance is negligible, calculate the indicated power. 6
- 10 a. With the help of T- S diagram, explain the effect of regeneration on Brayton cycle efficiency. 6
- b. Define volumetric efficiency as applied to compressors. Derive an expression for the same with usual notations. 8
- c. Which compression process needs minimum work input and which the maximum? Support your answer with the help of a P-V diagram. 6

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