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# P.E.S. College of Engineering, Mandya - 571401 <br> (An Autonomous Institution affiliated to VTU, Belagavi) <br> Third Semester, B.E. - Automobile Engineering <br> Semester End Examination; Dec - 2017/Jan - 2018 <br> Thermo Dynamics 

Time: 3 hrs
Max. Marks: 100
Note: Answer FIVE full questions, selecting ONE full question from each unit.

## UNIT - I

1 a. Define the following with examples:
i) Property
ii) Cycle
iii) Path function
iv) Quasi static process
v) Thermodynamics equilibrium
vi) Macroscopic approach.
b. Consider an alcohol and a mercury thermometer that read exactly $0^{\circ} \mathrm{C}$ at the ice point and $100^{\circ} \mathrm{C}$ at the stem point. The distance between the two points is divided into 100 equal parts in both thermometers. Do you think these thermometers will give exactly the same reading at a temperature of, say $60^{\circ} \mathrm{C}$ ? Explain.
c. Humans are most comfortable when the temperature is between $65^{\circ} \mathrm{F}$ and $75^{\circ} \mathrm{F}$. Express these temperature limits in ${ }^{\circ} \mathrm{C}$. Convert the size of this temperature range $\left(10^{\circ} \mathrm{F}\right)$ to $\mathrm{K},{ }^{\circ} \mathrm{C}$, and R .
2 a. Distinguish between heat and work in the thermodynamics.
b. Derive the expression for the polytropic work in a closed system.
c. In an air expression for the compressed air has an internal energy $450 \mathrm{~kJ} / \mathrm{kg}$ at the beginning of the expansion and an internal energy is $120 \mathrm{~kJ} / \mathrm{kg}$, calculate the heat flow to and from the cylinder.

## UNIT - II

3 a. State and explain the First law of the thermodynamics. Give its equation with reference to a cyclic and non-cyclic process.
b. $\quad 1.5 \mathrm{~kg}$ of liquid is stirred in a conducting chamber. During the process 1.7 kJ of heat are transferred from the liquid to the surroundings, while the temperature of the liquid is rising to $15^{\circ} \mathrm{C}$ Find $\Delta \mathrm{E}$ and W for the process.
c. Air flows steadily at the rate of $0.4 \mathrm{~kg} / \mathrm{s}$ through an air compressor, entering at $6 \mathrm{~m} / \mathrm{s}$ with a pressure of 1 bar and a specific volume of $0.85 \mathrm{~m}^{3} / \mathrm{kg}$, and leaving at $4.5 \mathrm{~m} / \mathrm{s}$ with a pressure of 6.9 bar and a specific volume of $0.16 \mathrm{~m}^{3} / \mathrm{kg}$. The specific internal energy of the air leaving $1 \mathrm{~s} 88 \mathrm{~kJ} / \mathrm{kg}$ greater than that of the air entering. Cooling water in a jacket surrounding the cylinder absorbs heat from the air at the rate of 59 kW . Calculate the power required to drive the compressor and the inlet and outlet pipe cross sectional areas.

4 a . What is a perpetual motion machine of first kind? Why is it impossible?
b. Apply steady flow energy equation to each of the following:
i) Boiler
ii) Nozzle
iii) Centrifugal pump
iv) Heat Exchanger.
c. A blower handles $1 \mathrm{~kg} / \mathrm{s}$ of air at $20^{\circ} \mathrm{C}$ and consumes a power of 15 kW . The inlet and outlet velocities of air are $100 \mathrm{~m} / \mathrm{s}$ and $150 \mathrm{~m} / \mathrm{s}$ respectively. Find the exit air temperature, assuming adiabatic conditions. Take Cp of air is $1.005 \mathrm{~kJ} / \mathrm{kg}-\mathrm{K}$.

## UNIT - III

5 a. What is thermal energy reservoir? Explain source and sink.
b. Define reversible engine. Show that of all the reversible heat engines working between any two constant but different thermal reservoir temperature, the reversible reversed heat engine will have the maximum COP.
c. A reversible heat engine operates between two reservoirs at $827^{\circ} \mathrm{C}$ and $27^{\circ} \mathrm{C}$. Engine drives a Carnot refrigerator maintaining $-13^{\circ} \mathrm{C}$ and rejecting heat to reservoir at $27^{\circ} \mathrm{C}$. Heat input to the engine is 2000 kJ and the net work available is 300 kJ . How much heat is transferred to refrigerant and total heat rejected to reservoir at $27^{\circ} \mathrm{C}$ ?
6 a. Define Kelvin-Plant statement and Clausius statement of $2^{\text {nd }}$ Law of thermodynamics and shown that they are equivalent.
b. Two reversible heat engines A and B are arranged in series, A rejecting heat directly to B. Engine a cold sink at a temperature of $4.4^{\circ} \mathrm{C}$, if the work output A is twice that of B , find;
i) The intermediate temperature between A and B
ii) The efficiency of each engine
iii) The heat rejected to the cold sink.

## UNIT - IV

a. With the help of P-V and T-S diagram, derive an expression for the standard efficiency of a semi-diesel cycle.
b. An Otto cycle operates between maximum and minimum pressure of 600 kPa and 100 kPa . The minimum and maximum temperatures in the cycle are $27^{\circ} \mathrm{C}$ and 1600 K . Determine thermal efficiency of cycle and also show it on T-S and P-V diagram.
8 a. What are the draw backs of a single stage compressor for producing high pressure? How are these overcome by multistage compression?
b. Derive an expression to determine the volumetric efficiency of a reciprocating compressor, considering the ambient and the inlet conditions of the compressors are same.
c. A reciprocating compressor of single stage and double acting type is running at 200 rpm with mechanical efficiency of $85 \%$. Air flows into compressor at the rate of $5 \mathrm{~m}^{3} / \mathrm{min}$ measured at atmospheric condition of $1.02 \mathrm{bar}, 27^{\circ} \mathrm{C}$. The compressed air is leaving at 8 bar following polytropic process with index of 1.3 . Compressor has clearance volume of $5 \%$ of stroke

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Determine the dimensions of cylinder, volumetric efficiency and power input required to drive the compressor if stroke to bore ratio is 1.5 .

UNIT - V
9 a. With neat sketches explain the working of vapour absorption refrigeration system.
b. What do you mean by refrigerant, refrigeration and refrigerator?
c. A refrigerator working on simple vapour compression cycle operates between the temperature of $25^{\circ} \mathrm{C}$ and $-15^{\circ} \mathrm{C}$ with NH 3 , refrigerant. Ammonia is found to be dry after compression and no under cooling of liquid refrigerant occurs in cycle. Calculate;
i) Refrigerating effect
ii) Mass flow rate per ton of refrigeration
iii) COP
iv) Power per TR
v) Represent on pH and TS diagram.

10 a. Define: i) Relative humidity
ii) Specific humidity
iii) Dew point temperature iv) Enthalpy of humid air v) Degree of saturation.
b. For the atmospheric air at room temperature of $30^{\circ} \mathrm{C}$ and relative humidity of $60 \%$ determine partial pressure of air, humidity ratio, dew point temperature, density and enthalpy of air.

