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P.E.S. College of Engineering, Mandya - 571 401

(An Autonomous Institution affiliated to VTU, Belagavi)

Fourth Semester, B. E. - Industrial and Production Engineering Semester End Examination; July / August - 2022 Material Science and Metallurgy

Time: 3 hrs Max. Marks: 100

Course Outcome's

The Students will be able to:

- CO1: Know the Fundamental concepts of Materials, and different Structures of Materials and common types of defects in the materials.
- CO2: Analyze the concept and mechanism of Fracture, Fatigue and Creep.
- Co3: Construct and analyze the different types of Solid Solutions and Iron Carbon Equilibrium diagram.
- CO4: Analyze the different heat treatment techniques to improve the specific properties of the engineering materials.
- CO5: Identify the composition, properties and application of ferrous, non-ferrous materials and composite materials.

<u>Note</u>: i) **PART-A** is compulsory. One question from each unit for maximum of 2 marks ii) **PART-B**: Answer any <u>TWO</u> sub questions (from a, b, c) from each unit for a Maximum of 18 marks.

Q. No.	Questions I : PART - A	Marks 10	BLs	COs
I a.	Define Fick's law of diffusion.	2	L1	CO1
b.	Write any two difference between normalizing and annealing process	2	L1	CO2
c.	Define composite material	2	L1	CO3
d.	Write the benefits of heat treatment.	2	L1	CO4
e.	Define unit cell.	2	L1	CO5
	II: PART - B	90		
	UNIT - I	18		
1 a.	What is APF? Calculate atomic packing factor for BCC and HCP structure	10	L3	CO1
b.	Briefly explain crystal imperfections	9	L1	CO1
c.	Explain stress strain diagram with all salient feature for mild steel	9	L1	CO1
	UNIT - II	18		
2 a.	With sketches explain stages of creep.	9	L1	CO2
b.	Define fatigue? explain SN curve	9	L1	CO2
c.	Briefly explain cooling curve for Cu-Ni material and also explain gib's phase	9	L1	CO2
	rule.	9	LI	CO2
	UNIT - III	18		
3 a.	Explain iron carbon diagram and also explain different phases of the same.	9	L1	CO3
b.	Explain TTT curve for hypo, hyper eutectoid steel.	9	L1	CO3
c.	With sketches explain microstructure of slowly cooled steel.	9	L1	CO3

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	UNIT - IV	18		
4 a.	Briefly explain with neat diagram tempering and martempering process.	9	L1	CO4
b.	Explain with heat diagram pack carbonizing	9	L1	CO4
c.	Explain properties, composition, and uses of gray CI and malleable cast Iron.	9	L1	CO4
	UNIT - V	18		
5 a.	Write the properties, composition, and uses of copper alloy, brasses, and bronzes.	9	L1	CO5
b.	Write the classification of composite material.	9	L1	CO5
c.	Write with neat diagram, fundamental production of MMC's	9	L1	CO5